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The reactants present in the
experiment are - $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$

- Molar mass = 147.0146

g/mol $\text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$ -Molar mass

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=184.2309 g/mol (b). The mixture is tested for an excess of ca view the full answer Previous question Next question

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dessaniel jaquez 10/18/17

experiment limiting reactant

co-workers: zach, sophie

objectives: to determine the limiting reactant in mixture of two soluble salts. . . .

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